## 3.139 The atmospheric reactivity of the NO3 radical.

Early Career Scientist

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## Abstract:

We describe the first direct measurements of the total reactivity of NO<sub>3</sub> in ambient air, in which cavity-ring-down spectroscopy is used to monitor the loss synthetically generated NO<sub>3</sub> after reacting with ambient trace-gases in a flow-tube. The instrument can measure NO<sub>3</sub> loss rate constants between 0.005 s<sup>-1</sup> and 45 s<sup>-1</sup> with an uncertainty of 16 % in the center of its dynamic range.

Results from the deployment of this instrument in a boreal forest in southern Finland and a rural mountain site in southern Germany are presented. In both cases,  $NO_3$  reactivity was driven by local meteorology coupled with biogenic emissions and displayed a strong vertical gradient with the highest reactivity measured below canopy level in the boreal forest. Very low  $NO_3$  reactivities were observed in the residual layer. Comparison of the measured  $NO_3$  reactivity with measurements of Volatile Organic Compounds (VOCs) indicated that the reactivity is dominated by reaction with monoterpenes, though a significant fraction of reactivity remained unattributed. During daytime, at both sites, more than 25% of the  $NO_3$  formed was removed via reaction with biogenic volatile organic compounds (BVOCs), implying a significant daytime loss of NOx and formation of organic nitrates and secondary organic aerosol via  $NO_3$  chemistry even though the nitrate radical is generally considered to be of importance at night.