3.128 Determination of the triple oxygen isotopic compositions of tropospheric carbon dioxide.

Early Career Scientist

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Abstract:

The triple oxygen isotopic compositions (Δ^{17} O) of tropospheric carbon dioxide (CO₂) can be a useful tracer to quantify carbon cycle in terrestrial environments. Traditionally, CO2 must be converted to O_2 to determine $\Delta^{17}O$ precisely. However, toxic and dangerous reagents such as BrF₅ must be needed for the reactions to convert CO₂ to O₂. Alternative safer, more simple, and more easy techniques should be needed for the routine measurements on the Δ^{17} O of tropospheric CO₂. In this study, we developed an alternative new method for the high precision measurements on the $\Delta^{17}{\rm O}$ of CO $_2$ using Cavity Ring-Down Spectroscopy (CRDS) for H₂O. First of all, atmospheric sample with CO₂ was introduced into pre-evacuated line and separated from atmospheric nitrogen (N_2) , oxygen (O_2) and water (H_2O) using the differences in the boiling points. Then, N_2O having the similar molecular weight and the similar boiling point with CO₂ was separated from CO₂ using a Porapak PS column under -70 °C. By reacting the purified CO₂ with H₂ at 600 °C under the existence of the nickel catalysts, CO2 was converted to methane (CH ₄) and water (H₂O). Subsequent to purify H₂O from CH₄, Δ^{17} O of H₂O was measured in CRDS. The new system developed in this study has several advantages over the conventional methods, such as (1) safe and easy operations, (2) less than 50 min for the single sample analysis, and (3) comparable precisions with previous methods. By using the new system, we can determine Δ^{17} O of tropospheric CO₂ with precision better than 0.015 % (1 σ). By using the new system developed in this study, we determined the difference in the $\Delta^{17}\text{O}$ values between tropospheric CO $_2$ in Nagoya and that in car exhaust, and found that tropospheric CO $_2$ in Nagoya was 0.22‰ higher in Δ^{17} O than that in car exhaust.