2.006 An inter-comparison of methods for HO2 and CH3O2 detection and kinetic study of the HO2 + CH3O2 cross-reaction in the Highly Instrumented Reactor for Atmospheric Chemistry (HIRAC).

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Abstract:

The hydroperoxy radical, HO₂, and methylperoxy radicals, CH₃O₂, participate in rapid chemical cycling at the heart of tropospheric oxidation. Laser-induced fluorescence (LIF) spectroscopy at low-pressure, known as the Fluorescence Assay by Gas Expansion (FAGE) technique, is most commonly used for the measurements of HO₂ in the atmosphere by conversion of HO₂ to OH by reaction with added NO followed by OH on-resonance LIF at 308 nm.¹ A new method has been developed for the sensitive and selective detection of CH₃O₂.² The method is similar to the FAGE method for HO₂ detection and consists in the titration of CH₃O₂ to CH₃O by reaction with NO followed by the detection of the resultant CH₃O by off-resonant LIF. Recently, the first near infrared CRDS measurements of HO₂ and CH₃O₂ in an atmospheric simulation chamber (HIRAC) were inter-compared against FAGE.^{3, 4} The good agreement between HO₂ and CH₃O₂, respectively concentrations measured using the two techniques at 1000 mbar of air provides a validation for the FAGE method for both HO₂ and CH₃O₂ detection.

The $HO_2 + CH_3O_2$ cross-reaction is important under clean, low NO_x levels, yet there are large uncertainties associated with its kinetics.⁵ The FAGE technique has been used to measure kinetic decays of HO_2 and CH_3O_2 radicals by the cross-reaction at 268, 295 and 323 K and 1000 mbar of air in the HIRAC chamber. The preliminary results have overlapping error limits at 1sigma level with the IUPAC preferred values at all three

temperatures.⁵

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